## University of Mumbai SYBSc/Semester-III/Chemistry Subject: - Chemistry-I Sample Questions

Date:-17/11/2020

## Multiple Choice Questions

1	Kohlrausch's law is not applicable for determination of			
	a) Degree of	b) EMF	c) Solubility	d) Ionization product
	dissociation		Product	of water
2	For spontaneity of reaction $\Delta G$ must be			
	a) Positive	b) Negative	c) Neutral	d) Zero
3	Sm <sup>-1</sup> is unit of			
	a) Conductance	b) Specific	c) Equivalent	d) Molar
		Conductance	Conductance	Conductance
4	The variation of equilibrium constant with temperature is given by equation.			
	a) Gibbs-Helmholtz	b) Vant's Hoff	c) Vant's Hoff	d) Gibbs free enegry
		isotherm	isochore	
5	If z -axis is molecular axis then hybridization for carbon atom in ethyne molecule is			
	a) sp $(s + px)$	b) sp (s + py)	c) sp (s + pz)	d) sp (s + pd)
6	Energy released when gaseous ions are brought from infinity to form a mole of solid			
	substance at O <sup>0</sup> K is known as			
	a) Ionization energy	b) Lattice energy	c) Dissociation	d) Enthalpy
			energy	
7	Limiting radius ratio 0.225-0.0.414 value indicates geometry.			
	a) Linear	b) Trigonal	c) Tetrahedral	d) Octahedral
8	Given reaction showsmechanism.			
	OCH <sub>3</sub> OCH <sub>3</sub> OCH <sub>3</sub>			
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	a) $S_N^1$ mechanism	b) $S_N^2$ mechanism	c) $S_N^i$ mechanism	d) Benzyne
				Mechanism
9	Epoxide react with HCN in presence of CaO it gives			
	a) 2-Amino ethanol	b) 2-Cyano ethanol	c) Halohydrin	d) Epoxide
10	Among following molecule which is having more acidic character?			
	a) o-methylphenol	b) p-methylphenol	c) Phenol	d) m-Nitrophenol